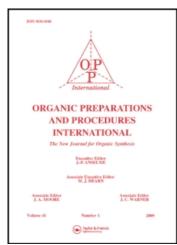
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PHASE-TRANSFER CATALYSED SYNTHESIS OF 4-BENZYLOXY-2-BUTANONES

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PHASE-TRANSFER CATALYSED SYNTHESIS OF 4-BENZYLOXY-2-BUTANONES

Submitted by G. Guillaumet, G. Coudert* and M. Mpassi (10/23/83)

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FRANCE

In connection with other synthetic work, we were interestted in preparing various 4-benzyloxy-2-butanones $(\underline{1})$, substituted on the aromatic ring. The only known compound $\underline{1a}$ was prepared by methods involving large excess of benzyl $alcohol^1$ and these methods could not be extended to the synthesis of compounds for which the requisite alcohols are either expensive or not readily available. We now report a general and efficient route to synthesize compounds $\underline{1}$ from methyl acetoacetate $\underline{2}$.

a)
$$Ar = C_6H_5$$
; b) $Ar = 2-MeOC_6H_4$; c) $Ar = 3-MeOC_6H_4$
d) $Ar = 4-MeOC_6H_4$; e) $Ar = 2,3-diMeOC_6H_3$
f) $Ar = 3,4-diMeOC_6H_3$; g) $Ar = 2-ClC_6H_4$

The keto ester $\underline{2}$ was first converted to the corresponding alcohol $\underline{4}$ by standard procedure. The benzyl chlorides were obtained quantitatively from the corresponding alcohols. Re-

actions of 4 with various benzyl chlorides were carried out by using liquid-liquid phase-transfer catalysis to yield ethers 5 in good yields. The reaction proceeded at 40° in a 50/50 mixture of benzene and aqueous sodium hydroxide, with 1.1 equivalent of chloride derivative and 0.1 equivalent of tetrabutyl-ammonium hydrogen sulfate. Careful hydrolysis of the acetal function was performed at room temperature in dioxane-water (50/50) in the presence of a catalytic amount of oxalic acid to yield the expected ketones 1 in high yields. A similar procedure could be performed without isolation and purification of the intermediary compounds 5.

EXPERIMENTAL SECTION

1,4-Dioxolanes (5).- Alcohol $\underline{4}$ (1 g, 7.5 mm), the appropriate benzyl chloride (8.25 mm) and tetrabutylammonium hydrogen sulfate (0.25 g, 0.75 mm) were dissolved in benzene (20 ml) and a 50% aqueous sodium hydroxide solution (20 ml) was then added. The reaction mixture was warmed at 40° with vigorous stirring for 6 hrs, cooled and poured into water. The aqueous phase was extracted with ether (4 x 25 ml). The organic phases were combined, washed with water, dried with magnesium sulfate and evaporated. The residual crude product $\underline{5}$ was purified by flash chromatography (silica gel column) before the following hydrolytic step or used directly. The yields of isolated products are based on alcohol $\underline{4}$.

 $\underline{5a}$, liquid (88% yield); IR neat: 1060 cm⁻¹ (C-O); NMR (CCl₄): δ 1.25 (s, 3 H, CH₃), 1.88 (t, 2 H, J = 7.3 Hz, CH₂), 3.29 (t, 2 H, J = 7.3 Hz, OCH₂), 3.77 (s, 4 H, OCH₂CH₂O), 4.42 (s, 2H,

 $CH_2Ar)$, 7.24 (s, 5 H, ArH).

6.60-7.55 (m, 4 H, ArH).

<u>Anal</u>. Calcd. for C₁₃H₁₈O₃: C, 70.27; H, 8.11.

Found: C, 70.10; H, 8.18.

 $\underline{5b}$, liquid (84% yield); IR (neat): 1060 cm⁻¹ (C-O); NMR (CCl₄): δ 1.26 (s, 3 H, CH₃), 1.92 (t, 2 H, J = 7.4 Hz, CH₂), 3.45-3.84 (m, 9 H, OCH₂, OCH₃ and OCH₂CH₂O), 4.51 (s, 2 H, CH₂Ar),

<u>Anal</u>. Calcd. for C₁₄^H₂₀O₄: C, 66.67; H, 7.93.

Found: C, 66.82; H, 8.02.

 $\underline{5c}$, liquid (84% yield); IR (neat): 1050 cm⁻¹ (C-O); NMR (CCl₄): δ 1.26 (s, 3 H, CH₃), 1.87 (t, 2 H, J = 7.2 Hz, CH₂), 3.44 (t, 2 H, J = 7.2 Hz, CH₂O), 3.69 and 3.75 (2 s, 7 H, OCH₃ and OCH₂-CH₂O), 4.34 (s, 2 H, CH₂Ar), 6.55-7.25 (m, 4 H, ArH).

Anal. Calcd. for $C_{14}^{H}_{20}^{O}_{4}$: C, 66.67; H, 7.93 Found: C, 66.51; H, 7.84.

 $\underline{5d}$, liquid (78% yield); IR (neat): 1055 cm⁻¹ (C-O); NMR (CCl₄): δ 1.25 (s, 3 H, CH₃), 1.88 (t, 2 H, J = 6.9 Hz, CH₂), 3.48 (t, 2 H, J = 6.9 Hz, CH₂), 3.48 (t, CH₂O), 3.70 and 3.79 (2 s, 7 H, OCH₃ and OCH₂CH₂O), 4.34 (s, 2 H, CH₂Ar), 6.78 (d, 2 H, J = 8.6 Hz, ArH), 7.20 (d, 2 H, J = 8.6 Hz, ArH).

Anal. Calcd. for $C_{14}^{H}_{20}^{O}_{4}$: C, 66.67; H, 7.93. Found: C, 66.83; H, 7.97.

 $\underline{5e}$, liquid (80% yield); IR (neat): 1065 cm⁻¹ (C-O); NMR (CCl₄): δ 1.26 (s, 3 H, CH₃), 1.88 (t, 2 H, J = 7.3 Hz, CH₂), 3.50 (t, 2 H, J = 7.3 Hz, CH₂O), 3.73 and 3.77 (2 s, 10 H, OCH₃ and OCH₂CH₂O), 4.42 (s, 2 H, CH₂Ar), 6.60-7.05 (m, 3 H, ArH).

<u>Anal</u>. Calcd. for $C_{15}^{H}_{22}^{O}_{5}$: C, 63.82; H, 7.80.

Found: C, 63.74; H, 7.77.

 $\underline{5f}$, liquid (77% yield); IR (neat): 1055 cm⁻¹ (C-O); NMR (CCl₄): δ 1.26 (s, 3 H, CH₃), 1.87 (t, 2 H, J = 7.0 Hz, CH₂), 3.45 (t, 2 H, J = 7.0 Hz, OCH₂), 3.6-3.9 (m, 10 H, OCH₃ and OCH₂CH₂O), 4.31 (s, 2 H, CH₂Ar), 6.6-6.8 (m, 3 H, ArH).

Anal. Calcd. for $C_{15}^{H}_{22}^{O}_{5}$: C, 63.82; H, 7.80. Found: C, 64.02; H, 7.91.

 $\underline{5g}$, liquid (85% yield); IR (neat): 1050 cm⁻¹ (C-O); NMR (CCl₄): δ 1.29 (s, 3 H, CH₃), 1.94 (t, 2 H, J = 7.2 Hz, CH₂), 3.60 (t, 2 H, J = 7.2 Hz, CH₂O), 3.80 (s, 4 H, OCH₂CH₂O), 4.51 (s, 2 H, CH₂Ar), 7.05-7.70 (m, 4 H, ArH).

<u>Anal.</u> Calcd. for C₁₃H₁₇ClO₃: C, 60.82; H, 6.63; Cl, 13.84. Found: C, 60.65; H, 6.80; Cl, 14.03.

4-Benzyloxy-2-butanones (1).- The crude acetal 5 was dissolved in a mixture of dioxane (10 ml) and water (10 ml) with a few milligrams of oxalic acid, and stirred at room temperature until the reaction was shown to be complete by tlc. The solution was extracted with ether (4 x 25 ml) and the combined extracts were washed with saturated sodium hydrogen carbonate (1 x 25 ml), dried over magnesium sulfate, filtered and evaporated in vacuo. The residual crude ketone 1 was applied to a silica gel column and the column was eluted with ether/petroleum ether (bp 45-65°). The yields of isolated products are based on alcohol 4.

<u>la</u>, liquid (85% yield); IR (neat): 1715 cm^{-1} (C=O); NMR (CCl₄): δ 1.98 (s, 3 H, CH₃), 2.52 (t, 2 H, J = 6.8 Hz, CH₂), 3.60 (t, 2 H, J = 6.8 Hz, CH₂O), 4.34 (s, 2 H, CH₂Ar), 7.24 (s, 5 H, ArH).

 $\underline{1b}$, liquid (79% yield); IR (neat): 1710 cm⁻¹ (C=O); NMR (CCl₄):

 δ 2.01 (s, 3 H, CH₃), 2.51 (t, 2 H, J = 6.5 Hz, CH₂), 3.62 (t, 2 H, J = 6.5 Hz, CH₂O), 3.70 (s, 3 H, OCH₃), 4.43 (s, 2 H, CH₂-Ar), 6.60-7.40 (m, 4 H, ArH).

<u>Anal</u>. Calcd. for C₁₂H₁₆O₃: C, 69.23; H, 7.69.

Found: C, 69.36; H, 7.78.

<u>lc</u>, liquid (81% yield); IR (neat): 1715 cm^{-1} (C=O); NMR (CCl₄); δ 2.0 (s, 3 H, CH₃), 2.51 (t, 2 H, J = 6.4 Hz, CH₂), 3.58 (t, 2 H, J = 6.4 Hz, CH₂O), 3.68 (s, 3 H, OCH₃), 4.36 (s, 2 H, CH₂-Ar), 6.65-7.40 (m, 4 H, ArH).

<u>Anal</u>. Calcd. for $C_{12}^{H}_{16}^{O}_{3}$: C, 69.23; H, 7.69.

Found: C, 69.41; H, 7.53.

<u>1d</u>, liquid (73% yield); IR (neat): 1715 cm^{-1} (C=O); NMR (CCl₄): δ 1.97 (s, 3 H, CH₃), 2.51 (t, 2 H, J = 6.4 Hz, CH₂), 3.57 (t, 2 H, J = 6.4 Hz, CH₂O), 3.64 (s, 3 H, OCH₃), 4.33 (s, 2 H, CH₂-Ar), 6.76 (d, 2 H, J = 8.8 Hz), 7.17 (d, 2 H, J = 8.8 Hz, ArH). Anal. Calcd. for $C_{12}H_{16}O_{3}$: C, 69.23; H, 7.69.

Found: C, 69.05; H, 7.52.

<u>le</u>, liquid (75% yield); IR (neat): 1710 cm^{-1} (C=O); NMR (CCl₄): δ 2.03 (s, 3 H, CH₃), 2.52 (t, 2 H, J = 6.5 Hz, CH₂), 3.62 (t, 2 H, J = 6.5 Hz, CH₂O), 3.73 and 3.75 (2 s, 6 H, OCH₃), 4.41 (s, 2 H, CH₂Ar), 6.60-7.10 (m, 3 H, ArH).

Anal. Calcd. for C13H18O4: C, 65.55; H, 7.56.

Found: C, 65.71; H, 7.44.

<u>lf</u>, liquid (71% yield); IR (neat): 1705 cm⁻¹ (C=O); NMR (CCl₄): δ 2.02 (s, 3 H, CH₃), 2.52 (t, 2 H, J = 6.3 Hz, CH₂), 3.56 (t, 2 H, J = 6.3 Hz, CH₂O), 3.70 and 3.74 (2 s, 6 H, OCH₃), 6.65-6.90 (m, 3 H, ArH).

Anal. Calcd. for C₁₃H₁₈O₄: C, 65.55; H, 7.56.

Found: C. 65.39; H, 7.47.

<u>lg</u>: liquid (81% yield); IR (neat): 1715 cm^{-1} (C=O); NMR (CCl₄): δ 2.07 (s, 3 H, CH₃), 2.59 (t, 2 H, J = 6.3 Hz, CH₂), 3.70 (t, 2 H, J = 6.3 Hz, CH₂O), 4.51 (s, 2 H, CH₂Ar), 7.05-7.65 (m, 4 H, ArH).

Anal. Calcd. for C₁₁H₁₃ClO₂: C, 62.12; H, 6.12; C1, 16.70. Found: C, 61.97; H, 6.03; C1, 16.86.

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